# organic papers

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#### **Key indicators**

Single-crystal X-ray study T = 120 KMean  $\sigma$ (C–C) = 0.002 Å R factor = 0.040 wR factor = 0.128 Data-to-parameter ratio = 13.2

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

## 2-(p-Tolyl)-1,3,2-benzodioxaborole

The title molecule, C<sub>13</sub>H<sub>11</sub>BO<sub>2</sub>, adopts a planar conformation and a stack/herringbone packing motif in the solid state.

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## Comment

Compound (I) was obtained via cobalt-mediated borylation of 4-iodotoluene, observed during our studies of the synthesis and reactivity of cobalt boryl complexes (Dai et al., 1996; Adams et al., 2006).



The asymmetric unit comprises one molecule (Fig. 1), which is nearly planar (r.m.s. deviation for all non-H atoms 0.057 Å), like its prototype 2-phenyl-1,3,2-benzodioxaborole (Zettler et al., 1974). The B atom is trigonal-planar; its coordination plane is inclined by  $2.9(1)^{\circ}$  to the catechol arene ring (i) and by 3.7 (1)° to the tolyl arene ring (ii). Molecules related via the btranslation form a stack with a mean interplanar separation of 3.52 (5) Å. Stacks are packed in a herringbone motif, in which planes of adjacent molecules are nearly perpendicular [dihedral angle 89.7  $(1)^{\circ}$ ].

## **Experimental**

To a stirred light-yellow solution of [Co(PMe<sub>3</sub>)<sub>3</sub>(BO<sub>2</sub>C<sub>6</sub>H<sub>4</sub>)<sub>2</sub>] (Dai et al., 1996) (0.110 g, 0.21 mmol) in hexane (2.0 ml), 4-iodotoluene (0.054 g, 0.25 mmol) was added at room temperature, resulting in a brown solution. After heating at 343 K overnight, the mixture became pink in colour. The solvent was then removed in vacuo and the residues were redissolved in THF (10 ml) to which was added excess CoCl<sub>2</sub>. The mixture was stirred for a further 15 min before being reduced to dryness in vacuo. The residues were then extracted with hexane and the resulting solution was concentrated in vacuo, during which a colourless solid appeared. This was redissolved by gentle heating, after which the solution was cooled slowly to give colourless crystals of (I) (0.015 g). <sup>11</sup>B NMR:  $\delta$  31.9. EI–MS m/z 210  $(M^{+}).$ 

#### Crystal data

$D_x = 1.293 \text{ Mg m}^{-3}$
Mo $K\alpha$ radiation
Cell parameters from 687
reflections
$\theta = 10.3 - 24.0^{\circ}$
$\mu = 0.09 \text{ mm}^{-1}$
T = 120 (2) K
Plate, colourless
$0.22\times0.15\times0.05$ mm

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## Data collection

Bruker SMART 6000 CCD areadetector diffractometer ω scans Absorption correction: none 9171 measured reflections 2486 independent reflections

#### Refinement

Refinement on $F^2$
$R[F^2 > 2\sigma(F^2)] = 0.040$
$wR(F^2) = 0.128$
S = 1.02
2486 reflections
189 parameters

Table 1

Selected geometric parameters (Å, °).

O1-C1	1.384 (2)	O2-B	1.393 (2)
O1-B	1.389 (2)	С7—В	1.533 (2)
O2-C2	1.384 (2)		
O1-B-O2	111.00 (14)	O2-B-C7	124.66 (13)
O1-B-C7	124.33 (14)		
O1-B-C7-C8	-3.0 (2)	O2-B-C7-C12	-3.4 (2)

1654 reflections with  $I > 2\sigma(I)$ 

All H-atom parameters refined  $w = 1/[\sigma^2(F_o^2) + (0.0681P)^2]$ where  $P = (F_o^2 + 2F_c^2)/3$ 

 $\begin{aligned} R_{\rm int} &= 0.071 \\ \theta_{\rm max} &= 27.5^\circ \end{aligned}$ 

 $h = -23 \rightarrow 17$ 

 $k = -6 \rightarrow 6$ 

 $l = -16 \rightarrow 16$ 

 $(\Delta/\sigma)_{\rm max} < 0.001$ 

 $\Delta \rho_{\rm max} = 0.25 \text{ e } \text{\AA}^{-3}$ 

 $\Delta \rho_{\rm min} = -0.19 \text{ e} \text{ Å}^{-3}$ 

All H atoms were refined isotropically, yielding the following distances:  $Csp^3 - H = 0.98$  (2) to 1.01 (2) Å and  $Csp^2 - H = 0.95$  (2) to 1.00 (2) Å.

Data collection: *SMART* (Bruker, 2001); cell refinement: *SMART*; data reduction: *SAINT* (Bruker, 2001); program(s) used to solve structure: *SHELXTL* (Bruker, 2001); program(s) used to refine



## Figure 1

Molecular structure of (I). Atomic displacement ellipsoids are drawn at the 50% probability level.

structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL*.

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